Convenient Synthesis of β -(1 \rightarrow 3)-Galactosyl Disaccharide α -Glycoside and Its Analogs as Mimic Units of Mucin-Type Carbohydrate¹

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 β -Galactosidase from porcine testes induced regioselective transglycosylation from lactose to the 3-position of 2-acetamido glycosides. When α -D-GalNAc-OC₆H₄NO₂-p was used as an acceptor, the enzyme synthesized mainly β -D-Gal-(1 \rightarrow 3)- α -D-GalNAc-OC₆H₄NO₂-p with its (1 \rightarrow 6) linked isomer. The use of an inclusion complex of the glycoside acceptor with β -CD increased the efficiency of transglycosylation by increasing the solubility of the acceptor. In the same way, the use of β -D-GalNAc-OC₆H₄NO₂-p as acceptor led to the preferential synthesis of β -D-Gal-(1 \rightarrow 3)- β -D-GalNAc-OC₆H₄NO₂-p over that of its (1 \rightarrow 6) linked isomer. β -D-Gal-(1 \rightarrow 3)- β -D-GlcNAc-OC₆H₄NO₂-p was also synthesized with β -D-GlcNAc-OC₆H₄NO₂-p acceptor by the consecutive use of β -D-galactosidases from porcine testes and *Bacillus circulans*. These enzyme reactions are efficient enough to allow the one-pot preparation of the desired disaccharide glycosides.

Key words: endo- α -N-actylgalactosaminidase, β -D-galactosidase, β -D-Gal- $(1 \rightarrow 3)$ - α -D-GalNAc-OC₆H₄NO₂-p, mucin-type oligosaccharide, transglycosylation.

The O-linked oligosaccharide components of glycoprotein are of interest because of their numerous roles in biological events such as cell-cell recognition (1). The common structure at the carbohydrate core region of mucin is a $3 \cdot O \cdot \beta$ -D-galactosyl-N-acetylgalactosamine unit α -glycosidically linked to serine or threonine. An organic chemical method for obtaining β -D-Gal- $(1 \rightarrow 3)$ - α -D-GalNAc- $OC_{6}H_{4}NO_{2}$ -p has been developed (2), but it involves various elaborate procedures for protection, glycosylation, and deprotection. From a practical viewpoint, the use of glycosidase is an attractive alternative for synthesis of such a disaccharide glycoside. On the other hand, Hedbys et al. have synthesized reducing disaccharides β -D-Gal-(1 \rightarrow 3)-D-GalNAc (3) and β -D-Gal-(1 \rightarrow 3)-D-GlcNAc (4) by transglycosylation utilizing β -D-galactosidase from bovine testes. Our interest was directed to an enzymic approach involving a $3 \cdot O \cdot \beta \cdot D$ -galactosyl-N-acetylgalactosamine unit α -glycosidically linked to p-nitrophenol as a mimic unit of mucin-type carbohydrate. This compound would be useful as an exogenous substrate for endo- α -N-acetylgalactosaminidase (endo- α -GalNAc-ase, EC 3.2.1.97) (5), as a probe for lectin and as a model compound of O-linked glycan. Thus, the object of the present investigation is to develop a system for selective transfer of a galactosyl residue to the C-3 position of an α -glycoside acceptor, based on this approach.

The present paper describes a preparative synthetic

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method for β -D-Gal- $(1\rightarrow 3)$ - α -D-GalNAc-OC₆H₄NO₂-p, which is a mimic unit of mucin type I, and its analogs, by means of porcine testes β -D-galactosidase-catalyzed transglycosylation using an inclusion complex of p-nitrophenyl glycoside acceptor with β -CD.

MATERIALS AND METHODS

Materials— β -D-Galactosidase [EC 3.2.1.23] from Bacillus circulans was obtained from Daiwa Kasei (Osaka). Endo- α -GalNAc-ase from Alcaligenes sp. was purchased from Seikagaku (Tokyo). β -D-Gal- $(1\rightarrow 4)$ - α -D-GlcNAc-OC₆H₄NO₂-p was synthesized by our method (6). Nagstatin, an inhibitor of N-acetyl- β -D-hexosaminidase (β -NAHase) (7), was obtained from Meiji Seika (Tokyo). All other chemicals were obtained from commercial sources.

Enzyme Assay— β -D-Galactosidase activity was assayed as follows. A mixture containing 2 mM o-nitrophenyl β -D-galactopyranoside in 0.9 ml of 50 mM sodium phosphate buffer (pH 6.0) and an appropriate amount of enzyme in a total volume of 0.1 ml was incubated for 10 min at 30°C. The reaction was stopped by adding 2 ml of 0.1 M Na₂CO₃, and then the liberated o-nitrophenol was determined spectrophotometrically at 420 nm. One unit of enzyme was defined as the amount hydrolyzing 1 μ mol of o-nitrophenyl β -D-galactopyranoside per min.

Partially Purified β -D-Galactosidase from Porcine Testes—A crude preparation of β -D-galactosidase was prepared by precipitation with ammonium sulfate from an extract of porcine testes by the same method as previously used with bovine testes (3). The crude β -D-galactosidase was used without further purification for the syntheses of β -D-Gal-(1 \rightarrow 3)- β -D-GalNAc-OC₆H₄NO₂-p and β -D-Gal-(1 \rightarrow 3)- β -D-GlcNAc-OC₆H₄NO₂-p. For synthesis of β -D-Gal-(1 \rightarrow 3)- α -D-GalNAc-OC₆H₄NO₂-p, the crude enzyme

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Abbreviations: endo- α -GalNAc-ase, endo- α -N-acetylgalactosaminidase; α -NAHase, N-acetyl- α -D-hexosaminidase; β -NAHase, N-acetyl- β -D-hexosaminidase; β -CD, β -cyclodextrin.

preparation was purified by removal of N-acetyl- α -D-hexosaminidase (α -NAHase) by column chromatography. The crude enzyme (1 g) was dissolved in 60 ml of 20 mM citrate phosphate buffer (pH 4.5) and charged onto a column $(\phi 3.2 \times 26 \text{ cm})$ of CM-Sepharose Fast Flow equilibrated with the same buffer. The column was washed with 200 ml of the same buffer. The enzyme was eluted with a linear gradient of concentration of NaCl from 0 (500 ml) to 1.0 M (500 ml) and the eluate was collected in 20-ml fractions. The fractions containing β -D-galactosidase activity (tubes 34-42) were eluted at about 0.5 M NaCl. Eluates of the corresponding enzyme fractions were combined and concentrated to small volume (1.5 ml) using an Amicon Diaflo unit. The concentrated enzyme solution was loaded onto a column ($\phi 2.6 \times 60$ cm) of HiLoad 26/60 Sephacryl S-200 equilibrated with 50 mM Tris-HCl buffer (pH 7.5) containing 0.5 M NaCl and 0.02% NaN₃. Elution was performed with the same buffer at a flow rate of 1.5 ml/min, and fractions of 2.5 ml were collected. The elution pattern showed two β -D-galactosidase peaks (F-1: tubes 32-38 and F-2: 44-56) and one α -NAHase peak (tubes 52-62). The fractions consisting the F-1 peak were completely devoid of α -NAHase, and these was combined and used for synthesis of β -D-Gal-(1 \rightarrow 3)- α -D-GalNAc-OC₆H₄NO₂-p (total activity: 0.12 U; specific activity: 0.044 U/mg).

Analytical Method—HPLC was performed with a YMCpacked column type AQ-312 (ODS) ($\phi 6 \times 150$ mm) and TSK-GEL G-Oligo-PW (\$\phi 7.8 \times 300 mm) in a Hitachi 6000-series liquid chromatograph equipped with an L-4000 ultraviolet detector. Elution of the former column was effected with 88:12 H₂O-CH₃OH, and that of the latter with H_2O . The flow rate was 0.8 ml/min at 40°C. ¹³C- and ¹H-NMR spectra were recorded on a JEOL JNM-EX 270. spectrometer. Chemical shifts are expressed in ppm relative to sodium 3-(trimethylsilyl)-propionate (TPS) as an external standard. FAB-MS analyses were carried out in the positive ion mode using a JEOL JMS DX-303HF mass spectrometer, coupled to a JEOL DA-800 mass data system. An accelerating voltage of 10 kV and a mass resolution of 1,000 was employed. A sample of 1 μ l in distilled water was loaded onto a probe tip and mixed with 1 μ l of glycerol as a matrix. Mass calibration was performed using Ultramark. Specific rotation was determined with a Digital Automatic Polarimeter PM-101 apparatus (Union Giken).

Preparation of β -D-Gal- $(1 \rightarrow 3)$ - α -D-GalNAc-OC₆H₄- NO_2 -p (A-1) and Its Positional Analog β -D-Gal-(1 \rightarrow 6)- α - $D \cdot GalNAc \cdot OC_{\rm fl}H_{\rm A}NO_{\rm 2} \cdot p (A \cdot 2) - \alpha \cdot D \cdot GalNAc \cdot OC_{\rm fl}H_{\rm A}NO_{\rm 2} \cdot$ p (120 mg) and β -CD (398 mg) were first dissolved in 18 ml of 20 mM sodium acetate buffer (pH 5.5), then lactose (2.48 g) was added, followed by partially purified β -D-galactosidase from porcine testes (0.12 U). The molar ratio of the donor to acceptor was about 20:1, and the total substrate concentration was about 14%. The mixture was incubated for 50 h at 40°C and the reaction was terminated by adding 8 ml of 1 M acetic acid followed by heating at 95°C for 10 min. Insoluble material was removed by centrifugation, and the supernatant was loaded onto a Toyopearl HW-40S column ($\phi 4.5 \times 90$ cm) equilibrated with 25% MeOH in aqueous solution, and the effluent solution was monitored by measuring the absorbance at 300 nm (p-nitrophenyl group) and 485 nm (carbohydrate content, determined by phenol-sulfuric acid method). As shown in Fig. 1, the chromatogram contained two peaks



Fig. 1. Chromatographic separation of transglycosylation products formed from lactose and α -D-GalNAc-OC₆H₄NO₂-*p* by porcine testes β -D-galactosidase. •, absorbance at 300 nm; \bigcirc , absorbance at 485 nm.

(F-1, tubes 56–61; and F-2, tubes 66–74) displaying coincident absorbance at 300 and 485 nm. These peaks were presumed to contain transfer products. Fraction F-2, after lyophilization to dryness followed by crystallization from ethanol, gave compound A-1 (27.2 mg). The eluates corresponding to F-1 were combined, concentrated and lyophilized to afford compound A-2 (8.8 mg). F-3 (tubes 93-104) contained α -D-GalNAc-OC₈H₄NO₂-p (64 mg) used as the acceptor.

The physical data for compound A-1 were identical to those of β -D-Gal-(1 \rightarrow 3)- α -D-GalNAc-OC₆H₄NO₂-p reported previously (2). NMR data (D₂O) of A-1: ¹H, δ 8.25 (d, 2H, J 9.2 Hz, m-Ph), 7.26 (d, 2H, J 9.2 Hz, o-Ph), 5.80 (d, 1H, < J 1 Hz, H-1), 4.53 (d, 1H, J 7.6 Hz, H-1'), 1.92 (s, 3H, NAc); ¹³C, δ 177.52 (C=O of Ac), 164.02 (Ph carbon attached to the phenolic oxygen), 145.12 (p-Ph), 128.79 (m-Ph), 119.39 (o-Ph), 107.64 (C-1'), 98.60 (C-1), 79.31 (C-3), 77.86 (C-5'), 75.29 (C-3'), 74.77 (C-5), 73.39 (C-2'), 71.37 (C-4,4'), 63.83 (C-6'), 63.74 (C-6), 50.10 (C-2), 24.73 (Me of Ac).

A-2 had: $[\alpha]_{b}^{25}$ +135.4°(c 0.5, H₂O); m.p. 207° (from ethanol); and m/z 505 (M+H)⁺. NMR data (Me₂SO- d_{6}): ¹H, δ 8.15 (d, 2H, J 9.2 Hz, m-Ph), 7.22 (d, 2H, J 9.2 Hz, o-Ph), 5.53 (d, 1H, J 3.5 Hz, H-1), 4.09 (d, 1H, J 7.6 Hz, H-1'), 1.84 (s, 3H, NAc); ¹³C, δ 163.09 (Ph carbon attached to the phenolic oxygen), 143.31 (p-Ph), 127.06 (m-Ph), 118.51 (o-Ph), 104.30 (C-1'), 97.70 (C-1), 76.03 (C-5'), 74.00 (C-3'), 72.10 (C-5), 71.59 (C-2'), 69.37 (C-4'), 69.37 (C-4), 69.02 (C-6'), 67.91 (C-3), 61.73 (C-6), 49.76 (C-2), 22.94 (Me of Ac).

Preparation of β -D-Gal- $(1 \rightarrow 3)$ - β -D-GalNAc-OC₆H₄NO₂p (B-1) and β -D-Gal- $(1 \rightarrow 6)$ - β -D-GalNAc-OC₆H₄NO₂-p (B-2)— β -D-GalNAc-OC₆H₄NO₂-p (66 mg) and β -CD (219 mg) were first dissolved in 12.5 ml of 20 mM acetate buffer (pH 5.5), then lactose (2.79 g) and Nagstatin (10 mM) were added, followed by the crude β -D-galactosidase (1.0 U) prepared by precipitation with ammonium sulfate from porcine testes. The molar ratio of the donor to acceptor was about 10 : 1, and the total substrate concentration was about 23%. The mixture was incubated for 12 h at 40°C and the reaction was terminated by heating at 95°C for 10 min. The resulting insoluble material was removed by centrifugation, and the supernatant was extracted with diethyl ether in order to remove p-nitrophenol liberated during the reaction. The aqueous phase was concentrated to small volume (20 ml) and loaded onto a Toyopearl HW-40S column as described above. The chromatogram showed two main peaks (F-1, tubes 67-75; and F-2, tubes 84-98) for which the absorbance at 300 nm coincided with that at 485 nm (data not shown). F-1 and F-2 were each concentrated and lyophilized to afford B-2 (2.8 mg) and B-1 (12.2 mg), respectively. F-3 peak (tubes 94-108), which showed absorbance only at 300 nm, contained β -D-GalNAc-OC₈H₄NO₂-p used as the acceptor.

NMR data (DMSO- d_6) of B-1: ¹H, δ 8.28 (d, 2H, J 8.9 Hz, m-Ph), 7.23 (d, 2H, J 9.2 Hz, o-Ph), 5.33 (d, 1H, J 8.6 Hz, H-1), 4.51 (d, 1H, J 7.3 Hz, H-1'), 2.00 (s, 3H, NAc); ¹³C, δ 171.79 (C=O of Ac), 163.79 (Ph carbon attached to the phenolic oxygen), 143.38 (p-Ph), 127.35 (m-Ph), 118.18 (o-Ph), 106.38 (C-1'), 99.89 (C-1), 80.39 (C-3), 77.38 (C-5'), 76.98 (C-5), 74.77 (C-3'), 72.22 (C-2'), 69.65 (C-4), 68.61 (C-4'), 62.07 (C-6'), 61.85 (C-6), 52.27 (C-2), 21.65 (Me of Ac). Other physical data for B-1 were identical to those of β -D-Gal-(1 \rightarrow 3)- β -D-GalNAc-OC₆H₄NO₂-preported previously (2).

B-2 had: $[\alpha]_{D}^{25} + 7.66^{\circ}$ (c 0.5, H₂0); and m/z 505 (M + H)⁺. NMR data (D₂O): ¹H, δ 8.28 (d, 2H, J 9.2 Hz, m-Ph), 7.28 (d, 2H, J 9.2 Hz, o-Ph), 5.31 (d, 1H, J 8.6 Hz, H-1), 4.48 (d, 1H, J 7.2 Hz, H-1'), 2.05 (s, 3H, NAc); ¹³C, δ 177.93 (C=O of Ac), 164.65 (Ph carbon attached to the phenolic oxygen), 145.44 (p-Ph), 128.93 (m-Ph), 119.47 (o-Ph), 106.16 (C-1'), 101.875 (C-1), 77.99 (C-5'), 77.32 (C-5), 75.51 (C-3'), 73.60 (C-2'), 73.22 (C-6), 71.84 (C-4), 71.42 (C-4'), 70.51 (C-3), 63.83 (C-6'), 54.91 (C-2), 24.96 (Me of Ac).

Preparation of β -D-Gal- $(1 \rightarrow 3)$ - β -D-GlcNAc-OC₆H₄NO₂p (B-3)—To a solution β -D-GlcNAc-OC₆H₄NO₂-p (1.0 g) and β -CD (3.3 g) in 50 ml of 20 mM acetate buffer (pH 5.5) were added lactose (21.0 g) and Nagstatin (0.1 mM), followed by the crude β -D-galactosidase from porcine testes (5 U). The molar ratio of the donor to acceptor was about 20 : 1, and the total substrate concentration was about 44%. The mixture was incubated for 50 h at 40°C and the reaction was terminated by heating at 95°C for 10 min. The resulting insoluble material was removed by centrifugation, and one half of the supernatant was loaded onto Chromatorex-ODS DM 1020T column ($\phi 3 \times 50$ cm) equilibrated with 20% MeOH in aqueous solution to remove the p-nitrophenol liberated during the reaction. The column was eluted with same solution. The eluate fractions (600 ml) showing coincident absorbance at 300 and 485 nm were combined and concentrated to a small volume (15 ml). The remaining half of the supernatant was similarly processed. The two batches of concentrated solution were combined and further applied to the Toyopearl HW-40S column described above. The elution conditions were the same as in Fig. 1. The eluate (20 ml fraction) showed two main peaks (F-1, tubes 50-57; and F-2, tubes 67-79) with coincident absorbance at 485 and 300 nm (data not shown). The latter peak was concentrated and lyophilized (133 mg), then treated with β -D-galactosidase from B. circulans in order to hydrolyze selectively β -D-Gal- $(1 \rightarrow 4)$ - β -D-GlcNAc-OC₆H₄- $NO_2 p$ (B-4) in the product mixture. Thus, F-2 (133 mg) was dissolved in 66.5 ml of 20 mM sodium phosphate buffer (pH 7.0) containing the β -D-galactosidase (40 U) and incubated at 40°C until the contaminant B-4 was no longer detected by HPLC; the required time for this reaction was 4 h on the average. The reaction was terminated by heating at 95°C for 10 min and the resulting insoluble material was centrifuged off. The supernatant was concentrated to a small volume (20 ml) and loaded onto the same column of Toyopearl HW-40S. F-2' fraction was eluted as a sharp peak (tubes 63-75). After concentration to dryness followed by crystallization from ethanol, this fraction gave compound B-3 in a yield of 75 mg. F-1 was concentrated to afford β -D-Gal-(1 \rightarrow 6)- β -D-GlcNAc-OC₆H₄NO₂-p (B-5, 40 mg) as reported previously (6). F-3 peak, showing absorbance only at 300 nm (tubes 85-100), contained β -D-GlcNAc-OC₆H₄NO₂-p used as an acceptor.

NMR data (D₂O) of B-3: ¹H, δ 8.22 (d, 2H, J 9.2 Hz, *m*-Ph), 7.18 (d, 2H, J 9.2 Hz, *o*-Ph), 5.34 (d, 1H, J 8.6 Hz, H-1), 4.47 (d, 1H, J 7.6 Hz, H-1'); ¹³C, δ 177.79 (C=O of Ac), 164.47 (Ph carbon attached to the phenolic oxygen), 145.51 (*p*-Ph), 128.91 (*m*-Ph), 119.37 (*o*-Ph), 106.34 (C-1'), 101.18 (C-1), 84.64 (C-3), 78.63 (C-5'), 78.15 (C-5), 75.33 (C-3'), 73.51 (C-2'), 71.36 (C-4), 71.23 (C-4'), 63.86 (C-6'), 63.29 (C-6), 57.16 (C-2), 24.98 (Me of Ac). Other physical data for compound B-3 were identical to those of β -D-Gal-(1 \rightarrow 3)- β -D-GlcNAc-OC₆H₄NO₂-*p* (8).

Enzyme Assay of Endo- α -GalNAc-ase from Alcaligenes sp.—The standard endo- α -GalNAc-ase assay was carried out as follows. A reaction mixture (1 ml) containing substrate (0.4 mM) and 0.19 mU of endo- α -GalNAc-ase in 20 mM citrate buffer (pH 6.0) was incubated at 40°C and samples (100 μ l) were taken at 2-min intervals during the reaction. After inactivation of each sample by adding 200 μ l of 1 M acetic acid, the liberated *p*-nitrophenol was determined by HPLC and spectrophotometry at 405 nm.

RESULTS AND DISCUSSION

Enzymic Synthesis of β -D-Gal- $(1 \rightarrow 3)$ - α -D-GalNAc-OC₆- $H_4NO_2 \cdot p$ (A-1)—The present paper describes a preparative synthetic method for A-1, a mimic unit of mucin type I carbohydrate, utilizing porcine testes β -D-galactosidasecatalyzed transglycosylation. The enzyme used in this case was completely devoid of α -NAHase activity, which degrades the acceptor substrate α -D-GalNAc-OC₆H₄NO₂-p (vide infra). On the other hand, α -D-GalNAc-OC₆H₄NO₂-p is only sparsingly soluble (0.1%) in aqueous medium. In general, the efficiency of the transglycosylation process is enhanced by the presence of a minimal amount of water and an excess of substrate (9). The problem of low solubility was partially solved by using β -CD, which is thought to form an inclusion complex with a *p*-nitrophenyl group, as reported in our previous study (10). Thus, the solubility of α -D-GalNAc-OC₆H₄NO₂-p was enhanced 6.5-fold by the presence of β -CD. When α -D-GalNAc-OC₆H₄NO₂-p was dissolved in amounts equimolar with β -CD, β -(1 \rightarrow 3) and β -(1 \rightarrow 6)-galactosyl disaccharide glycosides were observed by HPLC, in 22.1% total yield (based on the acceptor added) and in a ratio of 3 : 1. The maximum production of transfer products in the presence of β -CD was 2.5-fold higher (4.3 mM) than in its absence. This resulted in a significant improvement of the yield. Moreover, unreacted α -D-GalNAc-OC₆H₄NO₂-p acceptor, which is quite expensive, could be recovered by straightforward column chromatography and reutilized for synthesis.

Figure 2 shows a transglycosylation profile of the reaction with lactose and α -D-GalNAc-OC₆H₄NO₂-p in the absence and presence of β -CD. The maximum production of compound A-1 in the presence of β -CD was observed after 24 h, at which time it predominated 3-fold over A-2, and its concentration varied little during the subsequent reaction. On the other hand, the formation of A-2 was much slower, and maximum production was reached at ~ 50 h. In this way, the increased solubility of the acceptor by its formation of an inclusion complex with β -CD facilitated the production of the desired product A-1. We previously reported that the use of an organic co-solvent (50-60% Me₂SO, acetonitrile *etc.*) in transfer reactions utilizing glycosidases not only ensured a sufficient solubility of *p*-nitrophenyl glycosides, but also resulted in high yields of desired compounds (11-15). However, this concept was not adopted in the present reaction system because of the instability of the enzyme in organic co-solvent systems.

Enzymic Synthesis of β -D-Gal- $(1 \rightarrow 3)$ - β -D-GalNAc-OC₆- H_4NO_2 -p (B-1)—Crude β -D-galactosidase preparation from the extract of porcine testes was used directly for the preparation of B-1. Close attention was paid to the reaction system, because contaminant β -NAHase activity degrades β -D-GalNAc-OC₆H₄NO₂-p, the acceptor substrate, leading to loss of the transglycosylation. This problem was solved by using Nagstatin, a strong inhibitor of β -NAHase (7). When an inclusion complex of β -D-GalNAc-OC₆H₄NO₂-p with β -CD was used as the acceptor instead of α -D-GalNAc-OC₆H₄NO₂-p, two transfer products B-1 and B-2 were observed by HPLC in 15.3% total yield (based on the acceptor) and in a ratio of 5:1. These values are based on a time of 12 h for maximum yield of the desired β -(1 \rightarrow 3)-linked compound B-1. Figure 3 shows a transglycosylation profile of the reaction with lactose and β -D-GalNAc- $OC_6H_4NO_2$ -p. In this case, the transfer reaction led initially to the preferential formation of B-1 to B-2. However, once formation of B-1 reached its maximum, the amount decreased markedly during the subsequent reaction. B-2 formation was much slower, and the time for maximum production was \sim 240 h. At that time, its concentration surpassed that of B-1 in the latter stage of reaction. Thus, much more $(1 \rightarrow 3)$ -linked transfer product than the $(1 \rightarrow 3)$ -6)-linked product was formed in the initial stage of reaction. This result was in marked contrast to that with the α -



Fig. 2. Time course of porcine testes β -D-galactosidase-mediated isomer formation of A-1 and A-2 in the presence (closed symbols) and absence (open symbols) of β -CD. The amounts of A-1 (β 1 \rightarrow 3, \bullet , \Box) and A-2 (β 1 \rightarrow 6, \blacksquare , \Box) products as a function of time were examined on the 0.5-ml scale as described in the "MATE-RIALS AND METHODS" section, and samples were analyzed by HPLC during incubation.

D-GalNAc-OC₆ H_4NO_2 -*p* as an acceptor.

Enzymic Synthesis of β -D-Gal- $(1 \rightarrow 3)$ - β -D-GlcNAc-OC₆- $H_4NO_2 \cdot p$ (B-3)—The crude enzyme preparation was also used for the synthesis of B-3 with Nagstatin. When an inclusion complex of β -D-GlcNAc-OC₆H₄NO₂-p with β -CD was used as the acceptor, $\beta \cdot (1 \rightarrow 3) \cdot (B - 3)$, $\beta \cdot (1 \rightarrow 4) \cdot (B - 4)$, and $\beta \cdot (1 \rightarrow 6) \cdot (B \cdot 5)$ disaccharide glycosides were observed by HPLC, in 20.9% total yield (based on the acceptor) in the ratio of 48:20:32. These values are based on a time of 50 h for maximum production of the desired $\beta \cdot (1 \rightarrow 3)$ compound (B-3) (Fig. 4). The ratio of the three transfer compounds varied little during the entire course of reaction: the $(1 \rightarrow 4)$ -linked transfer product was formed in significant amounts along with the $(1 \rightarrow 3)$ -linked and $(1 \rightarrow 3)$ 6)-linked products. The regioselectivity is much lower than when the N-acetyl galactosaminide was used as acceptor. These transfer products were first separated into two fractions (F-1 and F-2) by chromatography on a Toyopearl HW-40S column. F-1 contained only compound B-5. The purification of F-2 into the desired compound B-3 was somewhat cumbersome, because this fraction was heavily contaminated by B-4. The B-4 was therefore selectively removed by hydrolytic treatment with B. circulans β -



Fig. 3. Time course of porcine testes β -D-galactosidase-mediated isomer formation of B-1 and B-2 in the presence of β -CD. The amounts of B-1 (β 1 \rightarrow 3, \bigcirc) and B-2 (β 1 \rightarrow 6, \bullet) products as a function of time were examined on the 0.5-ml scale as in Fig. 2.



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D-galactosidase, which does not hydrolyze B-3. When the lyophilized fraction was incubated at a 0.2% concentration with the enzyme, B-4 was hydrolyzed completely after 4 h. The hydrolysate was then rechromatographed on Toyopearl HW-40S, enabling the selective removal of B-4 from the F-2 fraction.

Substrate Specificity and Kinetics of Endo- α -GalNAcase—DiCioccio et al. have reported that endo- α -GalNAcase from Clostridium perfringens, which cleaves the endo-O-acetylgalactosaminidic linkage from serinyl or threoninyl oligosaccharide, hydrolyzes compound A-1 into β -D-Gal- $(1\rightarrow 3)$ -D-GalNAc and p-nitrophenol, that is, it can bypass a block of the disaccharide (5). In the present study, the substrate specificity of commercially available endo- α -GalNAc-ase from Alcaligenes sp. (16) was investigated using a series of synthetic p-nitrophenyl galactosyl-disaccharide glycosides (A-1, A-2, B-1, B-2, B-3, B-4, and B-5). The enzyme products were detected by HPLC using a TSK-GEL G-Oligo-PW column. The enzyme was capable of liberating a reducing disaccharide only from A-1. This indicates that its specificity for the substrate is very high.

By measuring the amount of *p*-nitrophenol as described in "MATERIALS AND METHODS," the initial velocities (v)were obtained directly from the initial slopes of the timecourse plots (2, 4, 6, and 8 min) of the reaction. Six different substrate concentrations (0.05-0.8 mM) were used per experiment. The kinetic parameters for A-1 of the enzyme assayed were calculated by the method of least squares with use of a [S]/v-[S] plot. The Michaelis constant (K_m) and k_0/K_m were 0.041 mM and 2,030 mM⁻¹. min⁻¹, respectively. This indicates that A-1 is a very sensitive substrate for the endo- α -GalNAc-ase from Alcaligenes sp. DiCioccio et al. have reported that the K_m value of A-1 for endo- α -GalNAc-ase from C. perfringens was 0.5 mM (5). The K_m value for this enzyme is much greater than that from Alcaligenes sp. Fan et al. have also reported that the activity of Diplococcus pneumoniae enzyme on DNS-Ser-GalNAc-Gal and DNS-Thr-GalNAc-Gal was lower than that of the *Alcaligenes* sp. enzyme (17). This indicates that the reactivity of endo- α -GalNAc-ases using these substrates depends greatly on the enzyme source.

In conclusion, β -D-galactosidase from porcine testes catalyzes formation of the β -(1 \rightarrow 3)-linked galactosyl disaccharide glycosides in a similar manner to that from bovine testes. The enzymic process for obtaining the desired compound A-1 is simple and the yield is sufficiently high to make the method practicable. From a practical viewpoint, a well-defined and homogenous substrate might facilitate the screening of new sources for endo- α -GalNAcase. Nitrophenyl glycoside can not only serve as an enzyme substrate, it can also be reduced to aminated glycosides. The amino function may be derivatized for reactions with electrophiles (18). The approach described above can be further extended to the synthesis of type 1 core (β -D-GlcNAc· $(1\rightarrow 3)$ · β ·D·Gal· $(1\rightarrow 3)$ ·D·GalNAc) and type 2 core $(\beta$ -D-Gal- $(1\rightarrow 3)$ - $[\beta$ -D-GlcNAc- $(1\rightarrow 6)$]-D-GalNAc) trisaccharides at the carbohydrate core region of mucin.

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